metal-organic papers

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Key indicators

Single-crystal X-ray study T = 299 KMean $\sigma(\text{C}-\text{C}) = 0.006 \text{ Å}$ R factor = 0.071 wR factor = 0.202 Data-to-parameter ratio = 12.7

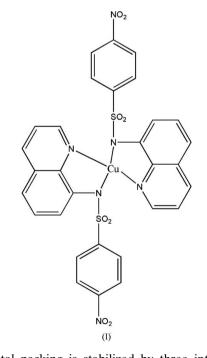
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Bis[4-nitro-*N*-(quinolin-8-yl)benzenesulfonamidato- $\kappa^2 N, N'$]copper(II)

The asymmetric unit of the title compound, $[Cu(C_{15}H_{10}N_3O_4S)_2]$, contains one half-molecule. The other half of the complex molecule is related by a C_2 axis running through the Cu atom. The Cu atom is four-coordinated by two quinoline N and two sulfonamide N atoms. In the crystal structure, the molecules are linked through intermolecular $C-H\cdots O$ hydrogen bonds.

Comment

Copper complexes based on sulfonamides have been investigated as potential reagents for the cleavage of nucleic acids (Macías *et al.*, 2003). Moreover, organic copper complexes based on quinoline rings can potently and selectively inhibit the chymotrypsin-like activity of the proteasome (Daniel *et al.*, 2004). In the light of this interest, we report here the structure of the title compound, (I). As in related structure reported by Macías *et al.* (2002), in (I), the Cu atom is four-coordinated by the two quinoline N and the two sulfonamide N atoms. A C_2 axis runs through the Cu atom. The Cu–N bonds are slightly longer to the quinoline N atom [1.996 (3) Å] than to the sulfonamide N atom [1.943 (3) Å]. Both bond lengths lie in the usual range. Selected bond distances and angles around the central Cu atom are given in Table 1.



The crystal packing is stabilized by three intermolecular $C-H\cdots O$ hydrogen bonds, which build a three-dimensional network, as shown in Fig. 2 and detailed in Table 2.

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Experimental

Compound (I) was prepared according to a literature procedure (Macías *et al.*, 2002). Single crystals of (I) suitable for X-ray data collection appeared after a few days from a methanol solution.

 $D_x = 1.567 \text{ Mg m}^{-3}$

Cell parameters from 25

Cu $K\alpha$ radiation

reflections

 $\theta = 4.1-22.8^{\circ}$ $\mu = 2.83 \text{ mm}^{-1}$

T = 299 (2) K

 $R_{\rm int} = 0.108$

 $\theta_{\text{max}} = 67.0^{\circ}$ $h = -16 \rightarrow 16$

 $k=-20\rightarrow 20$

3 standard reflections

frequency: 120 min

intensity decay: 5.0%

 $l = -15 \rightarrow 0$

Prism, dark green

 $0.38 \times 0.13 \times 0.10$ mm

Crystal data

 $\begin{bmatrix} Cu(C_{15}H_{10}N_3O_4S)_2 \end{bmatrix} \\ M_r = 720.18 \\ Monoclinic, C2/c \\ a = 14.101 (2) Å \\ b = 16.823 (2) Å \\ c = 13.172 (2) Å \\ \beta = 102.35 (1)^{\circ} \\ V = 3052.4 (7) Å^3 \\ Z = 4 \end{bmatrix}$

Data collection

Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (North *et al.*, 1968) $T_{min} = 0.556$, $T_{max} = 0.789$ 5582 measured reflections 2720 independent reflections 2074 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.071$ $wR(F^2) = 0.202$ S = 1.06 2720 reflections	$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.1447P)^{2}]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 1.00 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -1.58 \text{ e } \text{\AA}^{-3}$
	$\Delta \rho_{\rm min} = -1.58 \text{ e } \text{\AA}^{-3}$
	$\Delta \rho_{\min} = -1.58 \text{ e A}^{-5}$ Extinction correction: SHELXL97
214 parameters	
H-atom parameters constrained	Extinction coefficient: 0.0019 (3)

Table 1

Selected geometric parameters (Å, °).

1.943 (3)	Cu1-N1 ⁱ	1.943 (3)
1.996 (3)	Cu1-N2 ⁱ	1.996 (3)
162.1 (2)	N1-Cu1-N2 ⁱ	82.95 (14)
103.79 (14)	N2-Cu1-N2 ⁱ	136.37 (19)
	1.996 (3) 162.1 (2)	1.996 (3) $Cu1-N2^{i}$ 162.1 (2) $N1-Cu1-N2^{i}$

Symmetry code: (i) -x, y, $-z + \frac{1}{2}$.

Table 2

Hydrogen-bond	geometry	(A,	°).
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$D - H \cdots A$	$D-{\rm H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
C3-H3···O1 ⁱⁱ	0.93	2.50	3.354 (5)	153
$C11 - H11 \cdots O3^{iii}$	0.93	2.40	3.271 (6)	156
$C15-H15\cdotsO1^{i}$	0.93	2.54	3.335 (5)	144

Symmetry codes: (i) -x, y, $-z + \frac{1}{2}$; (ii) $-x - \frac{1}{2}$, $-y + \frac{1}{2}$, -z; (iii) x, -y + 1, $z - \frac{1}{2}$.

The H atoms were positioned with idealized geometry, with C–H = 0.93 Å, and were refined as riding with fixed isotropic displacement parameters set to 1.2 times U_{eq} of the parent atom. The residual electron-density peaks were located in the region of the Cu atom. The highest peak and deepest hole are 1.14 and 1.12 Å, respectively, from Cu.

Data collection: *CAD-4/PC* (Nonius, 1996); cell refinement: *CAD-4/PC*; data reduction: *REDU4* (Stoe & Cie, 1987); program(s) used to

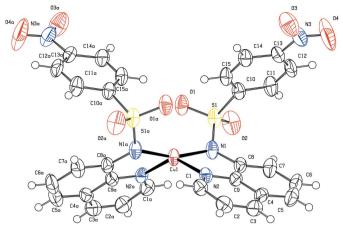
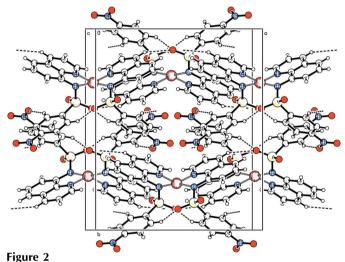


Figure 1

The molecular structure of (I), showing the atom labeling and displacement ellipsoids drawn at the 50% probability level. [Symmetry code: (a) -x, y, $\frac{1}{2} - z$.]



The molecular packing of (I), with hydrogen bonds shown as dashed lines.

solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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